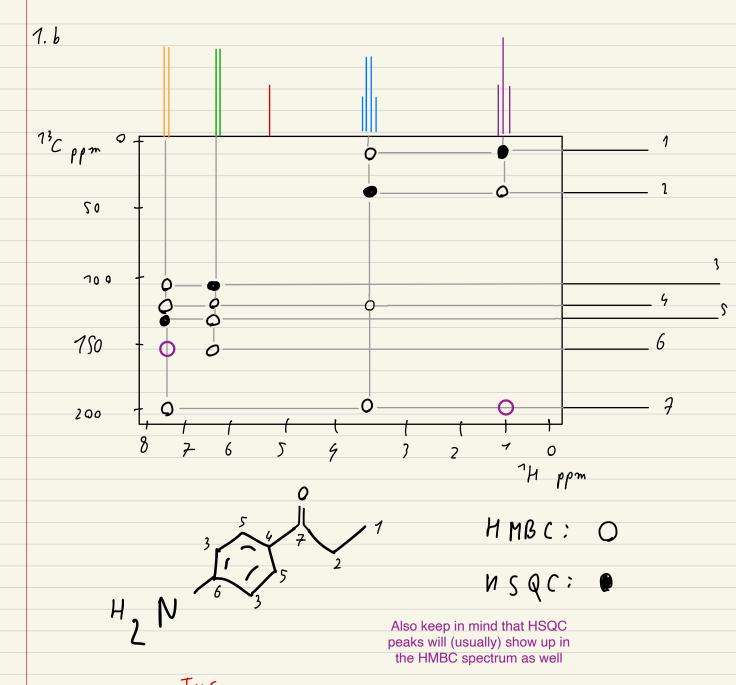
Excel	llent!

Great! Extra information: in solution state spectra, we don't usually see -NH2 or -OH protons because they are rapidly exchanging with the solvent. Also, proton peaks on aromatic rings usually have complex splitting patterns, so we just draw/describe them as multiplets



2. 
$$H_1$$

$$C = C$$

$$H_2$$

$$F_1$$

$$F_2$$

$$F_3$$

$$F_4$$

Q

The two hydrogen are chemically equivalent. However, because  $J_{H_1F_1} \neq J_{H_2F_1}$ , they are magnetically not equivalent.

 $H_3$  They are chemically equivalent but magnetically  $F_{n_1 n_4}$  rot equivalent because  $J_{n_1 n_4} \neq J_{n_2 n_4}$ .  $H_2$